

Reg No.: _____

Name: _____

APJ ABDUL KALAM TECHNOLOGICAL UNIVERSITY
B.Tech Degree S1,S2(S,FE) Examination May 2021 (2015 Scheme)

Course Code: BE101-06

Course Name: INTRODUCTION TO CHEMICAL ENGINEERING

Max. Marks: 100

Duration: 3 Hours

PART A

Answer all questions, each carries 3 marks

Marks

- | | | |
|---|--|-----|
| 1 | Write the stages involved in process design. | (3) |
| 2 | Define partial pressure and vapour pressure | (3) |
| 3 | List any three industrial applications of distillation. | (3) |
| 4 | Distinguish between conversion and yield of a chemical process. | (3) |
| 5 | Differentiate between block diagram and process flow diagram. | (3) |
| 6 | Explain the working principle of thermocouples. | (3) |
| 7 | Name any three air pollutants and its adverse effect in human body. | (3) |
| 8 | Write any three physical parameters for describing industrial waste water. | (3) |

PART B

Answer eight questions, (at least one full question from each module)

Module I

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|---|--|-----|
| 9 | Give an overview of chemical industries in Kerala. | (5) |
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Module II

- | | | |
|----|---|-----|
| 10 | Ethyl Chloride is produced by the gas phase reaction of HCl with Ethylene as
Follows $C_2H_4 + HCl \rightarrow C_2H_5Cl$
Find the mass flow rate of raw materials needed to make 50 kilograms per hour
Ethyl Chloride. | (5) |
|----|---|-----|

Module III

- | | | |
|----|---|-----|
| 11 | Differentiate evaporation and drying. | (5) |
| 12 | SO ₃ gas is obtained by the combustion of the FeS ₂ according to the reaction
$4FeS_2 + 15O_2 \rightarrow 2Fe_2O_3 + 8SO_3$
How many kilograms of pyrites are burned to obtain 200m ³ of Fe ₂ O ₃ ?
(specific gravity of Fe ₂ O ₃ =5.2) | (5) |

Module IV

- 13 Define rate of a reaction. What are the factors affecting rate of a reaction. (5)
14 Explain classification of chemical reactions. (5)

Module V

- 15 Draw P&ID symbols for any 5 instruments used in process piping. (5)
16 With a neat diagram explain principle and working of a venturi meter in flow measurement. (5)

Module VI

- 17 List any five novel materials and their uses. (5)
18 "Safety isn't expensive, it's priceless". Justify the statement as a chemical engineer. (5)

PART C

Answer six questions, (at least one full question from each module)

Module I

- 19 Explain various sectors of chemical industries. (6)

Module II

- 20 Write a note on mole%, mole fraction, mole ratio, weight %, weight fraction, weight ratio, volume %, volume fraction and volume ratio. (6)
21 Define the term equation of state. Write any three equations of state explaining the various terms involved. (6)

Module III

- 22 Explain the process of production of biodiesel. (6)

Module IV

- 23 Explain the laws of thermal radiation. (6)
24 With help of a diagram explain the features of CSTR and PFR. (6)

Module V

- 25 With a neat diagram explain working of a distillation column. Explain different control strategies that can be implemented to control the distillate quality. (6)

Module VI

- 26 What are the different challenges for chemical engineers in chemical plant safety? (6)
